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### CHE-551

# M.Sc. (Chemistry) IInd Year Examination Dec., 2023

## REACTION MECHANISM, PERICYCLIC REACTION, PHOTOCHEMISTRY STEREOCHEMISTRY

Time: 2 Hours] [Max. Marks: 70

Note: This paper is of Seventy (70) marks divided into two (02) Sections 'A' and 'B'. Attempt the questions contained in these Sections according to the detailed instructions given there in. Candidates should limit their answers to the questions on the given answer sheet. No additional (B) answer sheet will be issued.

#### Section-A

(Long Answer Type Questions) 2×19=38

Note: Section 'A' contains Five (05) Long-answer type questions of Nineteen (19) marks each.

Learners are required to answer any two (02) questions only.

**K-884** (1) P.T.O.

- Discuss the the photochemical cleavage of carbonyl compound with the help of the Norrish type-I and Norrish type-II.
- 2. Discuss the conformations of cyclohexane, 4-methyl cyclohexane and cyclohexanone
- 3. What is the carbene? Give the method of formation of carbene. Discuss the factors which affect the stability of carbene.
- 4. Discuss the mechanism of any *two* of the following :
  - (a) Baeyer-Villiger oxidation.
  - (b) Hoffmann Rearrangement.
  - (c) Wolf Rearrangement.
  - (d) Curtius Rearrangement.
- 5. What is Paterno-Buchi reaction? Discuss its mechanism along with the stereochemical consequences.

#### Section-B

## (Short Answer Type Questions) $4 \times 8 = 32$

**Note:** Section 'B' contains Eight (08) Short-answer type questions of Eight (08) marks each. Learners are required to answer any *four* (04) questions only.

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	(a)	Photochemistry of azo compounds.	
	(b)	[3, 3] sigmatropic rearrangement.	
4.	Explain ElcB mechanism with suitable example. How		
	is ElcB reaction differentiated from E2 reaction?		
5.	Give	e the mechanism of the any two reaction of	of the
	following:		
	(a)	Pinacol-pinacolane rearrangement	
	(b)	Ene reaction	
	(c)	Photo-smiles rearrangement	
6.	What is reactive intermediate? Explain the stability of		
	the Carbocation and cabanion.		
7.	Define cyclo addition reaction. What are [m + n]		
	cycloadditions? Explain with two examples.		
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1.

2.

3.

(a)

(b)

(c)

Discuss any two of the following:

Cis elemination in E2 reaction.

Draw the jablonski diagram. With the help of Jablonski

diagram describe, deactivation of excited states.

Hofmann rule.

Saytzeff rule.

Explain the following:

8. What is  $\pi$ - $\pi$  deg transition? Draw the HOMO and LUMO molecular orbital diagram of the 1, 3-butadiene and ethylene.

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